Real-time detection of airborne viruses on a mass-sensitive device

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We present real-time detection of airborne *Vaccinia* viruses using quartz crystal microbalance (QCM) in an integrated manner. *Vaccinia* viruses were aerosolized and neutralized using an electrospray aerosol generator, transported into the QCM chamber, and captured by a QCM crystal. The capture of the viruses on the QCM crystal resulted in frequency shifts proportional to the number of viruses. The capture rate varied linearly with the concentration of initial virus suspensions $(8.5 \times 10^8 - 8.5 \times 10^{10} \text{ particles/ml})$ at flow rates of 2.0 and 1.1 l/min. This work demonstrates the general potential of mass sensitive detection of nanoscale biological entities in air. © 2008 American Institute of Physics. [DOI: 10.1063/1.2956679]

Detection of nanoscale airborne organisms in a sensitive and rapid manner has been important for biological research, public health, and homeland security applications. Conventional bioaerosol detection techniques include culture-based analysis, microscope-based methods, and conventional polymerase chain reaction (PCR).³ However, these methods can require relatively long times for sample preparation, collection, and analysis.⁴ Recently, several techniques such as bioaerosol mass spectrometry (BAMS),⁵ real-time PCR,^{6,7} and flow cytometry, 8-10 have been developed. The BAMS is a rapid, real-time, and reagentless technique that can sample and analyze single cells, but it is most effective in the size range of 0.5–7 μm. Also, real-time PCR has demonstrated excellent capabilities of rapid sample identification and quantification. However, this technique requires the target analyte to be transferred to a fluid before analysis as well as the use of expensive reagents and significant expertise. Similarly, flow cytometry is very useful for rapid detection and quantification of microscale biological entities in fluids and, more recently, vibrating miniaturized channels have been reported to detect viruses and nanoparticles via their mass in fluids. 10

The quartz crystal microbalance (QCM) has been commonly used to detect a variety of nanoscale target analytes in liquid environment due to the advantages including simplicity of operation, real-time output, and label-free analysis. The basic QCM operation is described by the Sauerbrey equation where a mass change (Δm) in the crystal results in a subsequent change (Δf) in its resonance frequency.

$$\Delta m = -C_{\text{QCM}} \frac{\Delta f}{n}.$$
 (1)

Here, $C_{\rm QCM}$ is the mass sensitivity constant (17.7 ng cm⁻² Hz⁻¹ for the crystal used in this study) and n is the overtone number (n=3 in this study). ¹²

In this paper, we present real-time measurements of airborne *Vaccinia* viruses using a QCM. *Vaccinia* viruses are typically brick-shaped with an approximate size of 200 × 200 × 250 nm³. ^{13,14} Among numerous airborne microorganisms, *Vaccinia* virus was chosen for this study due to its robustness and resistance to environmental stresses. In fact because of these properties, it is regarded as a possible bioterrorism or biowarfare agent. ¹⁵ With this study, we demonstrate the potential of the QCM as a mass sensitive device to perform the detection of nanoscale entities in air. We use an electrospray aerosol generator as a means to generate the airborne virus particles.

Figure 1 shows a schematic of the overall experimental setup consisting of an electrospray aerosol generator (model 3480, TSI, Inc., St. Paul, MN) and a QCM (D300, Q-Sense, MD). Highly charged airborne *Vaccinia* viruses were gene-

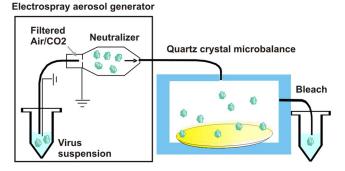


FIG. 1. (Color online) Schematic of airborne virus detection system by combining electrospray aerosol generation and QCM.

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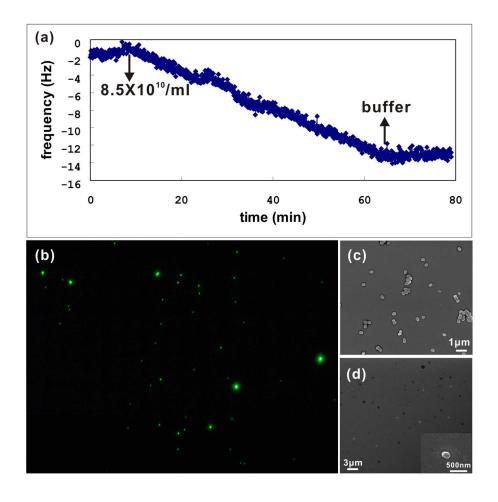


FIG. 2. (Color online) (a). Real-time response of QCM due to Vaccinia viruses $(8.5 \times 10^{10} / \text{ml})$ captured on a QCM crystal at a flow rate of 2.0 1/min. (b) Fluorescence micrograph $(370.0 \times 276.0 \ \mu m^2)$ of the sensor chip (crystal) confirming the capture of viruses. (c) A SEM image of Vaccinia viruses from a virus suspension before electrospray airborne particle generation. (d) A SEM image of Vaccinia viruses captured on a gold QCM crystal, where the white arrows indicate the viruses. The inset shows a single intact Vaccinia virus.

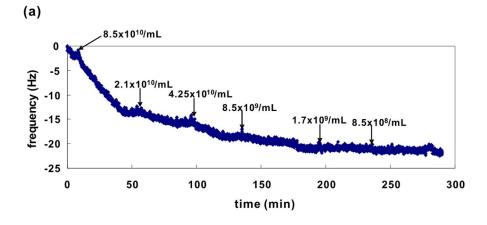
ated under a gas mixture of Air and CO2 using an electrospray aerosol generator from a virus suspension. The generated airborne viruses were then neutralized by a Polonium-210 source (Po-2042, NRD, Inc., Grand Island, NY), resulting in charge-equilibrated bioaerosols. These viruses were transported into the QCM chamber and adsorbed onto a gold-coated 5 MHz AT-cut quartz crystal (QSX 301, Q-Sense, MD), resulting in frequency shifts. A cleaned, goldcoated QCM crystal was mounted into the QCM chamber where constant temperature was maintained by a temperature controller.

Figure 2(a) shows the QCM response to injection of a suspension with a concentration 8.5 $\times 10^{10}$ particles/ml. As expected, the frequency decreases due to the adsorption of the virus particles on the crystal. The resulting 12 Hz frequency change corresponds 70.8 ng/cm² according to Eq (1). Given that the area of the QCM crystal is 0.7854 cm² and the mass of the single virus is approximately 9 fg, ^{13,14} the number of viruses can be approximated to be 6.16×10^6 . As an independent verification, we used fluorescence microscopy to count the viruses on the QCM chip. Figure 2(b) shows the fluorescent image of the captured viruses in an area of $370.0 \times 276.0 \ \mu \text{m}^2$. The viruses were stained by 3,3' dihexyloxacarbocyanine iodide [DiOC63], imaged under an epifluorescent microscope, and counted using IMAGEJ. The imaging/counting was performed on four different areas on the chip, yielding an average of 27478 viruses. Extrapolating this value to the entire area of the chip yields 21.1×10^6 viruses. Considering the possibility that other areas on the chip might very well have accumulated fewer number of viruses and that Sauerbrey equation is only an approximation here (since virus particles do not ex-Downloaded 10 Jul 2008 to 128.46.221.209. Redistribution subject to AIP license or copyright; see http://apl.aip.org/apl/copyright.jsp

actly constitute a rigid film), 12 the two independently estimated virus numbers are in reasonable agreement. Also shown in Fig. 2 are scanning electron microscope (SEM) images of Vaccinia viruses deposited onto a glass surface directly from a virus suspension (c) and viruses captured on the QCM crystal surface (d), indicating that viruses remain intact after the aerosolizing and capturing.

We also investigated the system's response to variation in the concentration of the virus suspension. Figure 3(a) shows a time response of the QCM to injections of different concentrations of aerosolized Vaccinia virus suspensions. The viruses were diluted with aqueous 10 mM ammonium acetate buffer (667404, Sigma-Aldrich). After stabilization with the buffer solution, six different concentrations of virus suspensions $(8.5 \times 10^8 - 8.5 \times 10^{10} \text{ particles/ml})$ were analyzed at a gas flow rate of 2.0 1/min. The buffer solution was also injected as a control between the subsequent virus injections.

Figure 3(b) demonstrates the dependence of virus capture rate on virus concentration, indicating an approximately linear relationship. Also, lowering the gas flow rate resulted in lower capture rates. It should be noted that the minimum detectable virus concentration can be approximated by considering the noise in the system, i.e., the capture rate signal caused by virus-free buffer injection. For 2.0 1/min, the noise level is approximately 0.0066 Hz/min, corresponding to 1.93×10^9 particles/ml, which compares well with the time response in Fig. 3(a) (the values were 0.0057 Hz/min and 5.77×10^9 /mL, respectively, for 1.1 1/min). The concentration of airborne viruses generated at the capillary tip of the



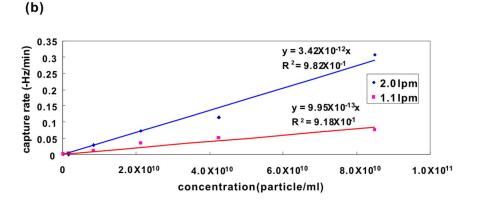


FIG. 3. (Color online) (a) Real-time measurements with different concentrations $(8.5\times10^8-8.5\times10^{10})$ particles/ml) of virus suspension where *Vaccinia* viruses were captured on a QCM crystal at a flow rate of $2.0\ 1/\text{min}$. (b) Variation of capture rate of *Vaccinia* viruses with different virus concentrations $(8.5\times10^8-8.5\times10^{10})$ particles/ml) at flow rates of $1.1\ \text{and}\ 2.0\ 1/\text{min}$. The lines indicate linear fits.

aerosol generation system can be approximated by dividing the product of the concentration of the starting virus suspension in fluid (viral particles/ml) and the suspension feed rate (432 nl/min) by the gas mixture flow rate (l/min). ¹⁶ Considering that only a small fraction of the particles (~9.5%) out of those generated at the capillary tip can reach the QCM crystal, ¹⁶ the minimum detectable airborne virus concentrations over the quartz crystal are calculated as around 40 and 210 particles/ml at a flow rate of 2.0 and 1.1 l/min, respectively, which are close to the detection limit (10 particles/ml) by BAMS. ¹⁷ The current detection limit can be further improved by higher flow rates in combination with secondary amplification techniques such as gold nanoparticle conjugation ¹⁸ and/or the use of electric fields to attract the viruses. ¹⁶

In conclusion, we presented the capture of airborne *Vaccinia* viruses in real-time using a mass sensitive device. We showed that the capture rate of airborne viruses measured by a QCM was linear with virus concentrations and that it had the potential for quantitative detection of airborne viruses. Our future efforts will be focused on extending this system to specific and selective detection of the airborne viruses. The system demonstrated here can also be used to study the impact of various aerosolized nanoscale entities on the environment and their interactions with living entities.

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¹Q. Y. Zeng, S. O. Westermark, A. Rasmuson-Lestander, and X. R. Wang, Appl. Environ. Microbiol. **70**, 7295 (2004).

²J. Kildeso and B. H. Nielsen, Ann. Occup. Hyg. **41**, 201 (1997).

³R. H. Williams, E. Ward, and H. A. McCartney, Appl. Environ. Microbiol. **67**, 2453 (2001).

⁴L. Birch, C. L. Archard, H. C. Parkes, and D. G. McDowell, Food Control. **12**, 535 (2001)

⁵H. J. Tobias, M. E. Pitesky, D. P. Fergenson, P. T. Steele, J. Horn, M. Frank, and E. E. Gard, J. Microbiol. Methods **67**, 56 (2006).

R. An, G. Mainelis, and L. White, Atmos. Environ. 40, 7924 (2006).
V. Pyankov, I. E. Agranovski, O. Pyankova, E. Mokhonova, V. Mokhonov, A. S. Safatov, and A. A. Khromykh, Environ. Microbiol. 9, 992 (2007).

⁸P. Chen and C. Li, Aerosol Sci. Technol. **39**, 231 (2005).

⁹P. Chen and C. Li, Analyst (Cambridge, U.K.) **132**, 14 (2007).

¹⁰T. P. Burg, M. Godin, S. M. Knudsen, W. Shen, G. Carlson, J. S. Foster, K. Babcock, and S. R. Manalis, Nature (London) 446, 1066 (2007).

¹¹S. M. Knudsen, J. Lee, A. D. Ellington, and C. A. Savran, J. Am. Chem. Soc. **127**, 2022 (2005).

Höök, A. Ray, B. Nordén, and B. Kasemo, Langmuir 17, 8305 (2001).
L. Johnson, A. K. Gupta, A. Ghafoor, D. Akin, and R. Bashir, Sens. Actuators B 115, 189 (2006).

¹⁴A. Gupta, D. Akin, and R. Bashir, Appl. Phys. Lett. **84**, 1976 (2004).

¹⁵I. E. Agranovski, A. S. Safatov, O. V. Pyankov, A. A. Sergeev, A. N. Sergeev, and S. A. Grinshpun, Aerosol Sci. Technol. 39, 912 (2005).

¹⁶J. Jang, D. Akin, K. S. Lim, S. Broyles, M. R. Ladisch, and R. Bashir, Sens. Actuators B **121**, 560 (2007).

¹⁷D. P. Fergenson, M. E. Pitesky, H. J. Tobias, P. T. Steele, G. A. Czerwieniec, S. C. Russell, C. B. Lebrilla, J. M. Horn, K. R. Coffee, A. Srivastava, S. P. Pillai, M.-T. P. Shih, H. L. Hall, A. J. Ramponi, J. T. Chang, R. G. Langlois, P. L. Estacio, R. T. Hadley, M. Frank, and E. E. Gard, Anal. Chem. 76, 373 (2004).

¹⁸W. A. Henne, D. D. Doorneweerd, J. Lee, P. S. Low, and C. A. Savran, Anal. Chem. **78**, 4880 (2006).